18%

Developed Due to Environmental Issues Chemical Fingerprinting of Materials

Fourth Conference on Aerospace Materials, Processes, and Environmental Technology Von Braun Center, Huntsville, AL September 20, 2000 Doris A. Smith, Ph.D. Technology Laboratories Lockheed Martin Space Systems Michoud Operations, New Orleans, LA (504) 257-0228 doris.smith@maf.nasa.gov

- Aerospace Materials
- Critical to performance
- Replaced or modified due to environmental restrictions
- Vary in composition from simple to complex; organic or inorganic; gas, liquid, or solid
- Subject to variations in composition due to formulation changes, ingredient substitutions, degradation, contamination, and mislabeling
- Must be adequately tested to detect variations



Building Blocks of Capabilities

Spectroscopy

Fourier Transform Infrared Spectroscopy (FTIR)

•Raman Spectroscopy •X-ray Fluorescence (XRF)

Inductively Coupled Plasma/Atomic Emission (ICP/AES)

Mass Spectrometry (MS)

Micro-Analysis

Scanning Electron Microscopy (SEM)
Energy Dispersive Spectrometry (EDS)

•Micro-FTIR •Micro-Raman

Chemical Microscopy

Surface Analysis

X-ray Photoelectron Spectroscopy (XPS)

Secondary Ion Mass Spectrometry (SIMS)

Gas Chromatography/Mass Spectrometry (GC/MS)
High Performance Liquid Chromatography (HPLC)

Gas Chromatography (GC)

Chromatography

•Gel Permeation Chromatography (GPC)

lon Chromatography (IC)

| Thermal Analysis

·Thermal Gravimetric Analysis (TGA)

Differential Scanning Calorimetry (DSC)



Used to identify and quantify elements present in samples

X-Ray Fluorescence

•Rapid detection of elements of atomic number ≥ 11

Solid & liquid samples, minimal sample preparation

Quantification requires matrix matched standards

Atomic Absorption

Rapid single-elemental quantitative analysis

·Sample must be in solution (accessory required for solids)

Sample preparation may be time consuming

Small linear response range, high matrix interference

Not applicable to most non-metals

inductively Coupled ମନ୍ଧଳଲା/ Atomic ଆନ୍ୟେତ୍ନ

·Multi-element qualitative and quantitative analysis

•Sample must be in solution (accessory required for solids)

Sample preparation may be time consuming

∘Large linear respons∈ range

Molecular Spectroscopy Spectroscopic Techniques: Used to identify and quantify molecular compounds present in samples

Spectroscopy Infrared

 Extensive reference libraries Minimal sample preparation Dipole moment change req. Molecular functional group identification • Complements Raman spec. ·Minor components masked by major Not applicable to aqueous samples Applicable to solids, liquids, gases

 Complements infrared spec. ·Molecular functional group identification

 Applicable to solids, liquids, aqueous ·Polarizability change required

Spectroscopy

Minimal sample preparation

Limited reference libraries

Not app. to colored or fluorescing samples

Spectrometry

 Widely applicable to volatile samples ·Organic compound identification

 Extensive reference libraries Chromatographic detector

Accessory required for non-volatile samples

Chromatographic Techniques:

- Used to separate and quantify components in samples
- Used in tandem with other techniques to identify components

Gas Chromatography (GC)

High Performance Liquid Chromatography (HPLC)

Gel Permeation Chromatography (GPC)

lon Chromatography (୮୯)

· Separation of volatile components within mixtures

Quantitative or qualitative analysis

Not applicable to thermally unstable components

·Not applicable to non-volatiles without derivatization

Separation of soluble components within mixtures

Quantitative or qualitative analysis

•Sample must be soluble in suitable solvent (many)

Method development time-consuming

Separation of components based on molecular size
Determination of molecular weight distribution

"Sample must be soluble in suitable solvent (few)

Separation and quantification of ionic species
Applicable to organic and inorganic

Method development time-consuming

Factors that Determine Fingerprinting Approach

Physical State:

Solid, liquid, or gas?

Homogeneous or distinct phases?

Sample size?

Chemical Properties: Single ii

s: Single ingredient or complex mixture?

Major, minor, or trace components?

Organic, inorganic, or combination?

Masking of one component by another?

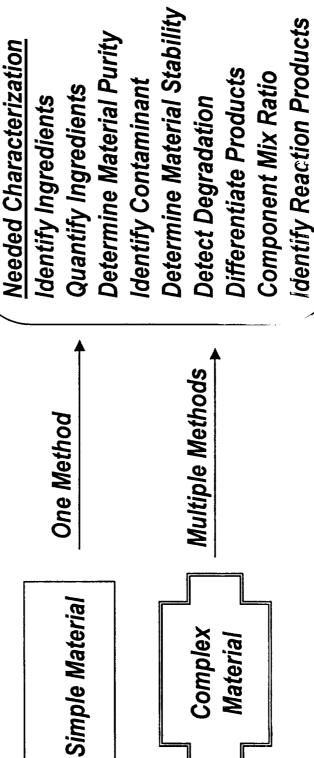
Separation of components required?

-- Information Required: Qualitative or quantitative data?

Bulk or surfac⊜ composition?

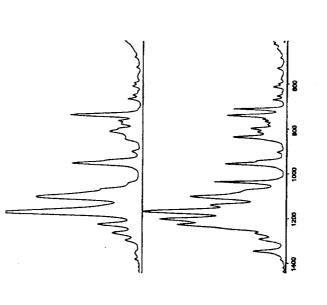
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Fingerprinting: Combination of instrumental analysis methods that diagnostically characterize a material



Simple Materials and Approach: HCFC-225 and HCFC-225G

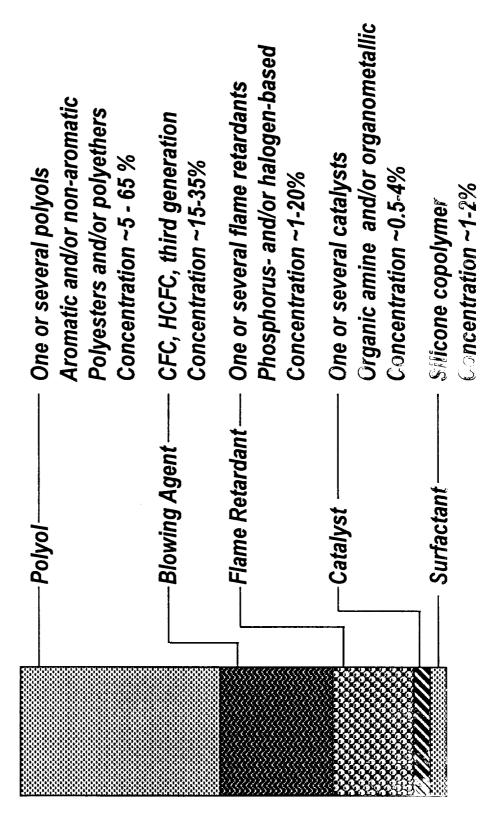
HCFC-225G Single Isomer Trace Impurities HCFC-225 Two isomers Trace Impurities



FTIR: Rapid Differen€ation

GC: Isomer Ratio, % Purity

Complex Material Example: Urethane Foam Component



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Complex Material Approach: Urethane Foam Component

ICP/AES: Tin, lead concentration (organometallic) Polyol conc. based on volatile portion Vo∉atile amine catalyst concentration Halogen-based flame retardant conc. Phosphorus-based flame ret. conc. Rapid blowing agent identification Detectable polyol concentration Concentration of polyol diluent Functional groups, polyol type Molecular weight distribution Blowing agent concentration Blowing agent degradation Phosphorus concentration ICP/AES: Sifeon concentration Flame Retardant - ICP/AES: GC/MS: HPLC: HPLC: GPC: FTIR: FTIR: GC: Blowing Agent — GC: Surfactant Catalyst

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Complex Material and Approach: Phenolic Resin

Ctarting Materials	Pnenolic Resin	
Stalling Materials	Phenol	Methv
Phenol	Cormoldobiodo	
Formaldehyde	rormaidenyde	
Colone	Solvent	Trime
Solveill	Catalvst	Dinuc
Catalyst	Water	Polvn

Methylol Phenols
Dimethylol Phenols
Trimethylol Phenol
Dinuclear Phenols
Polynuclear Phenols
uclear Pl ynuclear

Phenol and methylol phenol concentrations HPLC:

Solvent concentrations

Molecular weight distribution (resin advancement) GPC:

Functional groups (resin advancement) FTIR:

Concentration of metal due to metal hydroxide catalysts ICP/AES:

Ammonium hydroxide catalyst concentration

Formaldehyde and water concentrations

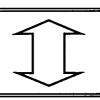
Supplier Partnership is Vital Element in Fingerprint Program

Material Suppliers:

- Provide information on formulation and chemistry
- Supply samples of formulation ingredients
- Avoid changes to material formulation when possible
- Notify us of necessary changes to material

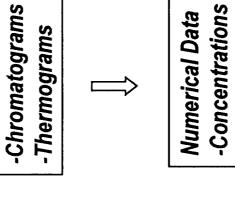


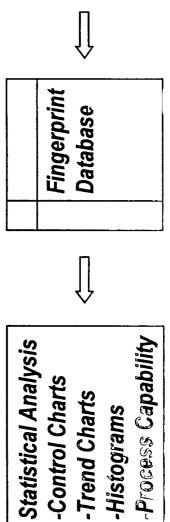
- Use information to understand material's chemistry and threats to availability
- Use information and samples to develop fingerprint methods
- Supply fingerprint data
- Safeguard proprietary information

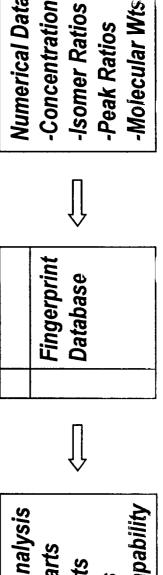


Fingerprint Databases & Trending to Detect Material Variations





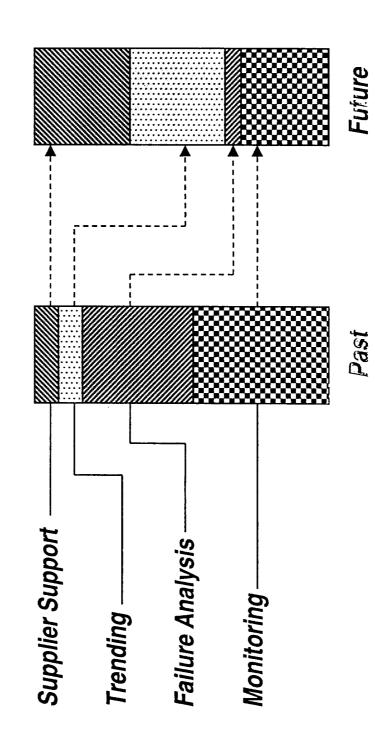


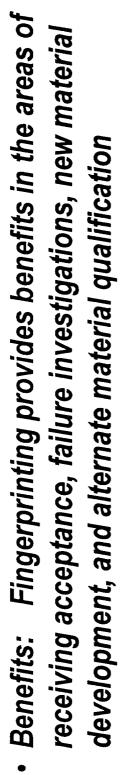


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Chemical Fingerprinting

Evolving Role of Fingerprinting in Aerospace Industry





- Multipurpose methods with diagnostic capability
- Quantitative databases and reference libraries
- Increased material reliability
- Ensured future replication of successful materials
- Expeditious problem resolution
- Automated sample analysis
- Reduced cost of material requalifications
- Increased supplier communication

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